



P-014-1042003

Seat No. _____

M. P. M. (Sem. II) (CBCS) Examination

July - 2018

BP202T : Pharmaceutical Organic Chemistry - I

Faculty Code : 014

Subject Code : 1042003

Time : 3 Hours]

[Total Marks : 75

1 Answer the following questions : 20

- (1) Define organic compound. Classify it with examples.
- (2) Define and classify isomers.
- (3) Explain Sp^3 Hybridization of Alkane.
- (4) What is Dienes ? Classify it with example.
- (5) Explain Saytzeffs orientation for addition reaction of alkene.
- (6) Justify the statement : Bromination of alkane is more selective than chlorination.
- (7) Write reaction involved in Benzoin and Aldol condensation.
- (8) Explain method for synthesis of Grignard reagent.
- (9) Comment : Order for basicity of amine is $2^\circ > 1^\circ > 3^\circ$ in aqueous solution.
- (10) What is Diel-Alder reaction ?

2 Answer the following questions : (Any Two) 20

- (1) What is Alkyl halide ? Explain reaction, mechanism and stereochemistry of nucleophilic aliphatic substitution reaction in Alkyl halide.
- (2) Define Carbonyl compounds. Give any two methods for preparation and reactions of aldehyde and ketone.
- (3) Give structure and uses of following compounds :
 - (a) Trichloroethylene
 - (b) Propylene glycol
 - (c) Chloral hydrate
 - (d) Salicylic acid
 - (e) Ethylenediamine

3 Answer the following questions : (Any Seven)

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- (1) Write a note on structural isomerism in organic compound.
 - (2) Enumerate chemical reactions of alkane and explain halogenations in detail.
 - (3) Explain E_2 versus E_1 reactions in alkene.
 - (4) Write name of qualitative tests for identification of carboxylic acid, ester, amide, alcohol and aliphatic amine.
 - (5) Explain structure of alcohol and explain laboratory methods for preparation of alcohol.
 - (6) Write reaction and mechanism of Cannizzaro reaction and Perkin condensation.
 - (7) Explain stability and free radical addition reactions of conjugated diene.
 - (8) Explain Acidity of carboxylic acids and effect of substituents on acidity of carboxylic acid.
 - (9) Write structure of Tartaric acid, Vanilin, Glycerol, Hexamine and Amphetamine.
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